



## **Explosive Bonding of Refractory Metal Liners**

**by William S. de Rosset**

**ARL-TR-3267**

**August 2004**

## **NOTICES**

### **Disclaimers**

The findings in this report are not to be construed as an official Department of the Army position unless so designated by other authorized documents.

Citation of manufacturer's or trade names does not constitute an official endorsement or approval of the use thereof.

Destroy this report when it is no longer needed. Do not return it to the originator.

# **Army Research Laboratory**

Aberdeen Proving Ground, MD 21005-5069

---

**ARL-TR-3267****August 2004**

---

## **Explosive Bonding of Refractory Metal Liners**

**William S. de Rosset**

**Weapons and Materials Research Directorate, ARL**

REPORT DOCUMENTATION PAGE			Form Approved OMB No. 0704-0188		
<p>Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing the collection information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing the burden, to Department of Defense, Washington Headquarters Services, Directorate for Information Operations and Reports (0704-0188), 1215 Jefferson Davis Highway, Suite 1204, Arlington, VA 22202-4302. Respondents should be aware that notwithstanding any other provision of law, no person shall be subject to any penalty for failing to comply with a collection of information if it does not display a currently valid OMB control number.</p> <p><b>PLEASE DO NOT RETURN YOUR FORM TO THE ABOVE ADDRESS.</b></p>					
1. REPORT DATE (DD-MM-YYYY) August 2004		2. REPORT TYPE Final		3. DATES COVERED (From - To) January 2004–July 2004	
4. TITLE AND SUBTITLE Explosive Bonding of Refractory Metal Liners			5a. CONTRACT NUMBER		
			5b. GRANT NUMBER		
			5c. PROGRAM ELEMENT NUMBER		
6. AUTHOR(S) William S. de Rosset			5d. PROJECT NUMBER AH84		
			5e. TASK NUMBER		
			5f. WORK UNIT NUMBER		
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) U.S. Army Research Laboratory ATTN: AMSRD-ARL-WM-MB Aberdeen Proving Ground, MD 21005-5069			8. PERFORMING ORGANIZATION REPORT NUMBER ARL-TR-3267		
9. SPONSORING/MONITORING AGENCY NAME(S) AND ADDRESS(ES)			10. SPONSOR/MONITOR'S ACRONYM(S)		
			11. SPONSOR/MONITOR'S REPORT NUMBER(S)		
12. DISTRIBUTION/AVAILABILITY STATEMENT Approved for public release; distribution is unlimited.					
13. SUPPLEMENTARY NOTES					
14. ABSTRACT <p>TPL, Inc. has successfully bonded a pure tantalum liner to the inside of an M242 Bushmaster barrel using a low-detonation-velocity explosive. This report examines the governing equations of the explosive bonding process as they apply to this particular situation. The relevant properties of other higher strength tantalum alloys are examined to see if they are suitable for explosive bonding, with the expectation that they would be better able to resist the wear forces at the rifled bore surface. For all candidate materials, attention is paid to the values of critical impact pressure, the critical flow transition velocity, the critical angle for jet formation, and maximum flyer plate velocity. Example plots are provided to indicate the bounds of explosive welding parameters that will result in a good bond.</p>					
15. SUBJECT TERMS explosive bonding, gun tube liners, refractory metals, wear and erosion, Bushmaster					
16. SECURITY CLASSIFICATION OF:			17. LIMITATION OF ABSTRACT  UL	18. NUMBER OF PAGES  42	19a. NAME OF RESPONSIBLE PERSON William S. de Rosset
a. REPORT UNCLASSIFIED	b. ABSTRACT UNCLASSIFIED	c. THIS PAGE UNCLASSIFIED			19b. TELEPHONE NUMBER (Include area code) (410) 306-0816

---

## Contents

---

<b>List of Figures</b>	<b>v</b>
<b>List of Tables</b>	<b>v</b>
<b>Acknowledgments</b>	<b>vii</b>
<b>1. Introduction</b>	<b>1</b>
<b>2. Background</b>	<b>2</b>
<b>3. Governing Equations</b>	<b>5</b>
<b>4. Application of Governing Equations</b>	<b>8</b>
<b>5. Discussion</b>	<b>12</b>
<b>6. Summary</b>	<b>14</b>
<b>7. References</b>	<b>15</b>
<b>Distribution List</b>	<b>17</b>

INTENTIONALLY LEFT BLANK.

---

## List of Figures

---

Figure 1. Donor tube wall thickness as a function of the donor tube outer radius for a fixed final liner thickness. ....	3
Figure 2. Standoff normalized to donor tube wall thickness as a function of donor tube outer radius. ....	4
Figure 3. Strain imparted to cylinder as a function of donor tube outer radius. ....	5
Figure 4. Geometry of explosive bonding setup. ....	6
Figure 5. Relation of pressure P in MPa vs. $\alpha$ in degrees for $V_c = 5$ km/s. ....	10
Figure 6. Bounding plots for explosive bonding of tantalum and two alloys. ....	11
Figure 7. Bounding plots for niobium. ....	12
Figure 8. Relation between $\alpha$ (in degrees) and standoff for a specific case. ....	14

---

## List of Tables

---

Table 1. Tantalum alloy material properties. ....	9
Table 2. Tantalum material properties. ....	9
Table 3. Properties of niobium. ....	12

INTENTIONALLY LEFT BLANK.



---

## **Acknowledgments**

---

The author is indebted to Brian Scott for providing background information on explosive bonding. He also recognizes the help of Steven Segletes for advice in writing the report. Dattatraya P. Dandekar was the source of the equation-of-state data. Discussions with Jonathan Montgomery and Bob Lowey (TPL, Inc.) were helpful in writing the background section of the report. Finally, the author is indebted to Larry Burton for his critical review of the report.

INTENTIONALLY LEFT BLANK.

---

## 1. Introduction

---

Gun barrel wear and erosion has been a major determinant in the useful life of most Army guns. It is a problem that has been addressed with propellant additives, barrel coatings, and reduced flame-temperature propellants. The problem has become more of an issue with the push to use higher-performance, higher flame-temperature propellants. Thus, the need to reduce barrel wear and erosion is established on prolonging the life of current gun barrels as well as providing the opportunity to introduce new propellants into future gun systems.

As part of a Small Business Innovative Research (SBIR) Phase 2 program, TPL, Inc. has demonstrated its ability to line the inside of both a rifled and smooth-bore (honed out) M242 Bushmaster 25-mm gun barrel with pure tantalum (Lowey, 2002) using an explosive bonding process. Initial firing tests at the U.S. Army Aberdeen Test Center showed that the tantalum-lined barrels have a remarkable resistance to wear and the explosive bonding process produced an extremely strong bond between the tantalum and gun steel. Most of the work was conducted with a smooth-bore gun, and the question of wear on a rifled surface was still an issue.

This successful effort has prompted continuance of the work through a Manufacturing Technology (Mantech) program. The goal of the program is to reduce wear and erosion of the M242 25-mm Bushmaster gun tube. One of the tasks of this program is to examine the choice of liner materials to see if another candidate material is more suitable. It may be that pure tantalum is too soft to withstand the forces exerted on the lands and grooves of rifled guns. Tantalum is also considered an expensive material, and since cost avoidance is one of the principal tenets of the Mantech program, high material cost is an issue.

One criterion for choice of a liner material is its cladability. This report investigates the material properties that make materials amenable to being explosively bonded to gun steel. In the next section, the work that TPL did to bond the tantalum liner to the M242 will be reviewed.

Particular attention will be paid to the characteristics of the explosive and the geometry of the tantalum cylinder used in the TPL work. The governing equations used in explosive bonding are presented in section 3. These equations indicate what material parameters are important for the process to be successful. (Note that other important characteristics of the liner, such as hardness, resistance to chemical attack, and machinability, are not addressed in this particular task.)

Section 4 takes the values of those material parameters for tantalum and several of its alloys to calculate the collision angle and collision velocity that give a good explosive bond. Results are also presented for niobium. The final section discusses the suitability of these metals and the implications for the choice of explosive.

---

## 2. Background

---

There is a certain amount of trial and error in establishing the operating parameters used in explosive bonding. This is evident in the number of different explosive formulation variations used by TPL in its work with the M242 Bushmaster gun barrel (Lowey, 2002). The final report for the Phase 2 effort omitted several important details; therefore, some suppositions about this work will be made for the remainder of this report. For instance, the actual formulation used to clad the three M242s was not explicitly specified in the final report; however, it can be inferred from the report that the explosive had a detonation velocity between 1.7 and 2.2 km/s.

The characteristic velocity, known as the Gurney velocity,  $\sqrt{2E}$ , can be estimated from the following:

$$\sqrt{2E} = D/2.97, \quad (1)$$

where D is the detonation velocity (Cooper, 1996). For this specific case,  $\sqrt{2E}$  is calculated to be 0.74 km/s, using the upper limit of D (2.2 km/s). Values of the Gurney velocity for standard military explosives generally fall between 2 and 3 km/s (Zukas and Walters, 1998).

There were many steps needed to process the M242 gun tube. The parts dealing primarily with the explosive bonding process itself were filling a tantalum donor tube with explosive, inserting the donor tube into the gun tube, centering the donor tube so that it was concentric with the gun tube, evacuating the gun tube, and detonating the explosive. The gun tube had been bored out to an inner diameter of 1.064 in (27.03 mm) to accommodate the liner thickness. (The diameter reported in Lowey [2002] was given as 1.0547 in. This is a typographical error [Lowey, 2004b].) The original dimensions of the tantalum donor tube were 0.75 in (19.05 mm) for the diameter and 0.065 in (1.651 mm) for the wall thickness. Given the density of the explosive as 0.66 g/cm<sup>3</sup> (Lowey, 2004a) and the density of tantalum as 16.65 g/cm<sup>3</sup>, the charge-to-mass ratio (C/M) can be calculated as 0.0856. Using the Gurney equation for the metal velocity V of the expanding cylindrical wall, we get the following:

$$V = \sqrt{2E} (M/C + 1/2)^{-1/2} = 0.212 \text{ km/s}. \quad (2)$$

The tantalum donor tube used in the explosive bonding to the M242 gun tube underwent a large amount of plastic strain. The radial component of strain  $\epsilon$  on the outer surface of the cylinder can be found using the values of the following initial and final cylinder outer diameters:

$$\begin{aligned} \epsilon &= \ln (\text{final cylinder diameter}/\text{initial cylinder diameter}) \\ &= \ln (1.064/0.75) = 35\%. \end{aligned} \quad (3)$$

There is a certain amount of plastic strain associated with the explosive bond itself. Consequently, an important material characteristic for explosive bonding is material ductility. However, in this particular instance, the need for material ductility is increased due to the fact

that the tantalum donor tube has to expand and stay together within the gun tube. TPL made an attempt to use tantalum-10% tungsten (Ta-10W) as the liner material (Lowey, 2002). No successful bonds were ever achieved. One possible cause cited for this failure was that the Ta-10W might have had too many interstitial impurities. Also, the starting material may have been too strain-hardened. Mulligan et al. (2002) point out that "... commercially pure electron beam melted tantalum has an elongation of ~30 to 40% and a yield strength of ~5 ksi at 1000 °C, while Ta-10W has an elongation of ~25% (within limits of explosive bonding)..." If the same initial donor tube dimensions were used for Ta-10W as were used for the pure tantalum, the allowable elongation limit may have been exceeded.

The need to have liner ductility in excess of 30% elongation may be reduced if the initial donor tube diameter is increased. (The tantalum tubes used by TPL, Inc. were purchased as off-the-shelf items and therefore came in a standard size.) However, there is a limit to the diameter size of the donor tubes. This is due to the standoff (initial distance between the outer donor tube wall and the inner gun wall) needed in the explosive bonding process. A rough "rule of thumb" is that the standoff should be between 1/2 and 1× the flyer plate thickness (Wylie et al., 1970). In any event, the cylinder wall thickness can be calculated as a function of the outer radius of the cylinder, assuming that the M242 is honed to the same dimensions (1.064-in inner diameter) and that the same final liner thickness (0.044 in) is achieved. This relation is shown in figure 1.

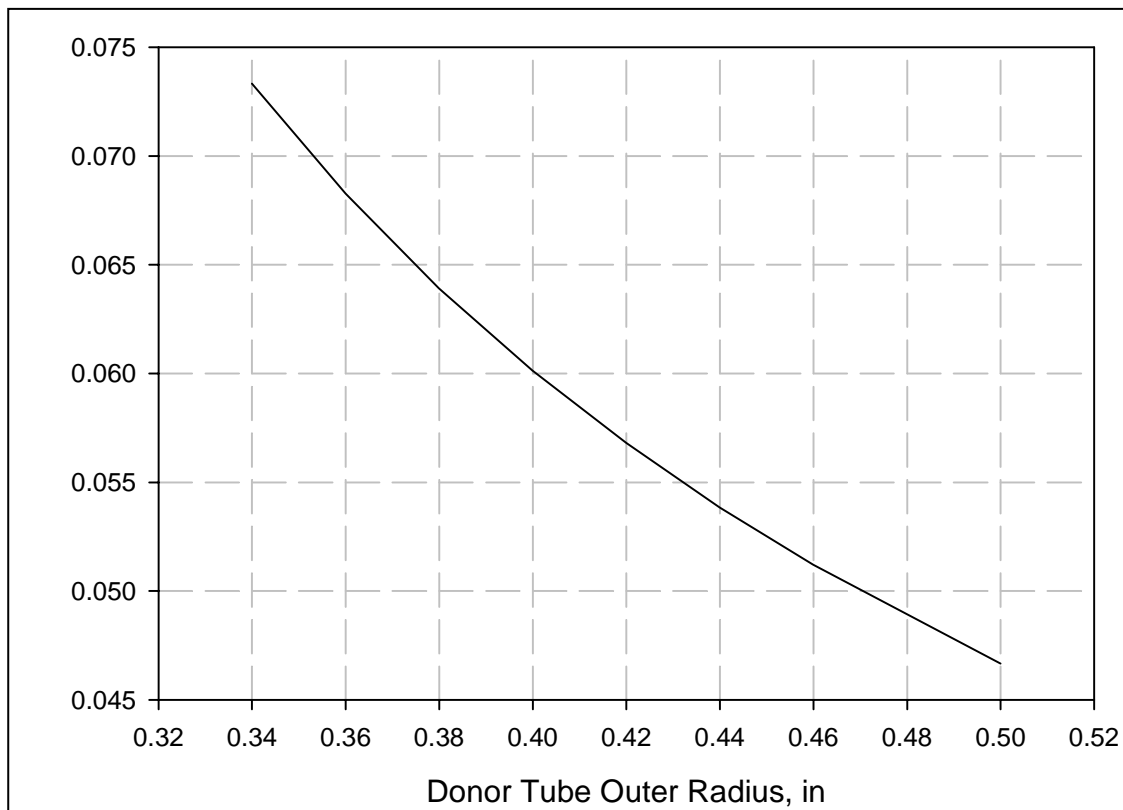


Figure 1. Donor tube wall thickness as a function of the donor tube outer radius for a fixed final liner thickness.

As the donor tube outer radius changes, the standoff (distance from the outer diameter of the donor tube to the inner diameter of the gun tube) normalized to the donor tube wall thickness changes. This is shown in figure 2. Finally, the amount of strain imparted to the donor tube can be calculated as a function of the donor tube outer radius. This is shown in figure 3.

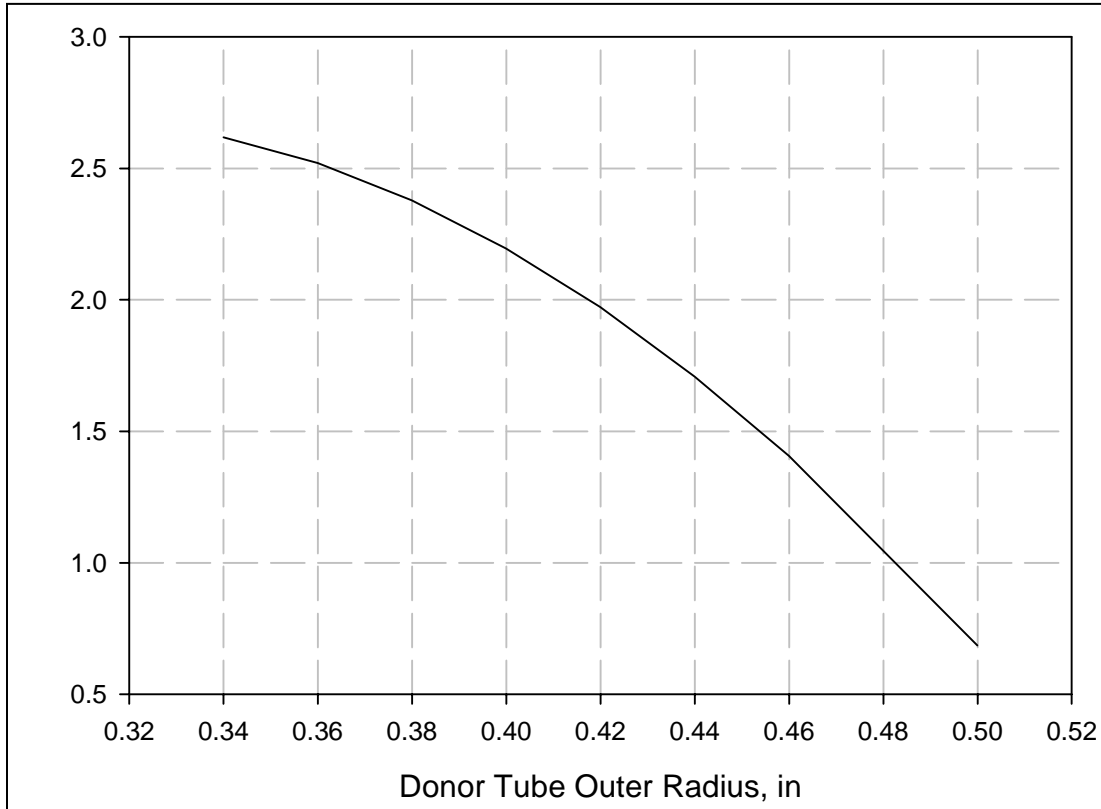


Figure 2. Standoff normalized to donor tube wall thickness as a function of donor tube outer radius.

These figures indicate that there is a range of possible initial conditions. The exact choice of these parameters will depend upon many other factors, including type of explosive used, availability of liner material with specified dimensions, and allowable elongation of the liner material. Similar plots can be made for other gun systems or for different choices of final liner thickness or initial gun tube diameter. The main point of these three figures is that if a normalized standoff less than 2 can be used, a lower elongation requirement can be met. For instance, a normalized standoff of 1.5 will result in a final elongation of less than 20%.

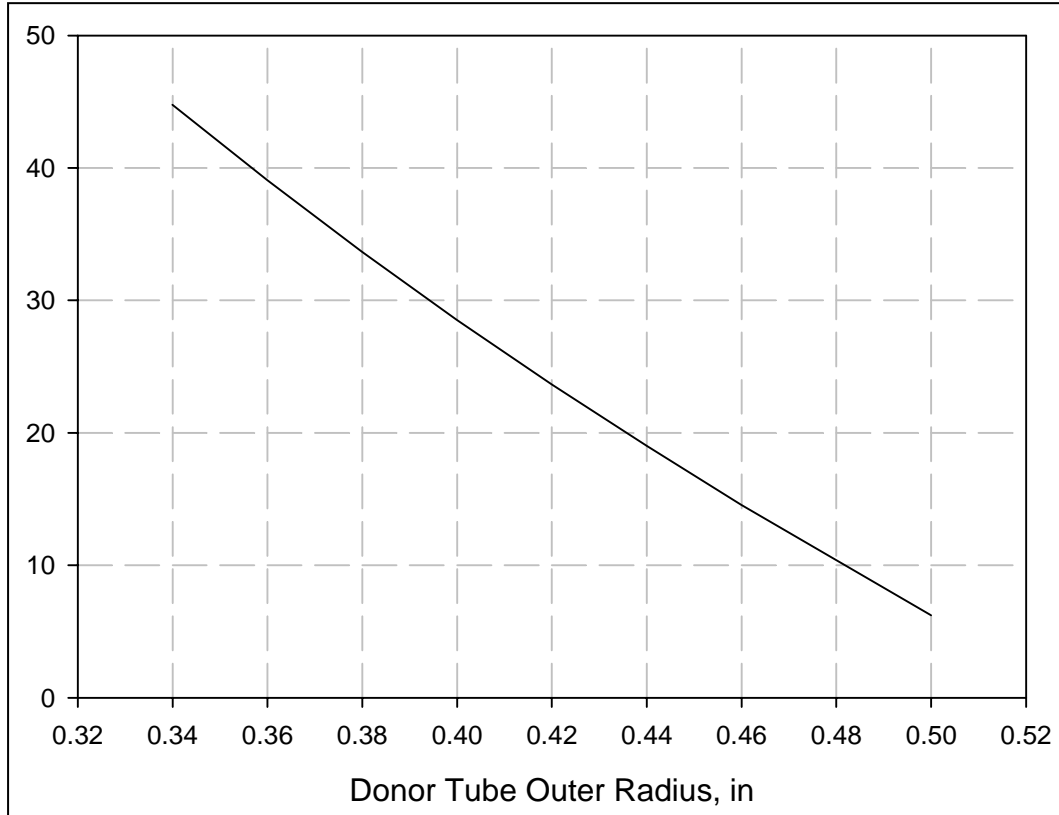


Figure 3. Strain imparted to cylinder as a function of donor tube outer radius.

### 3. Governing Equations

Explosion bonding has been used commercially for over 40 years to weld dissimilar metals that are otherwise difficult to join. The technology is relatively mature, and the governing equations have been documented in many publications. Carpenter and Wittman (1975) provide an excellent review of the technology, and most of what is presented in this section is taken from their work. (See this reference for a more thorough discussion of the equations.) In particular, they present four boundary conditions necessary to provide optimum explosion bonding characteristics. These are the critical angles for jet formation, the critical impact pressure, the critical flow transition velocity, and a maximum impact velocity.

Figure 4 briefly describes these governing equations and their rationale and shows the geometry of the explosive bonding setup. A constant standoff geometry is used for bonding the metal liner to the gun tube wall. (We distinguish between the initial liner configuration, called the donor tube, and the gun barrel.) In this figure,  $V$  is the flyer plate velocity,  $D$  is the velocity of detonation of the explosive,  $V_c$  is the collision point velocity, and  $\alpha$  is the angle between the donor tube and the gun barrel at the collision point.

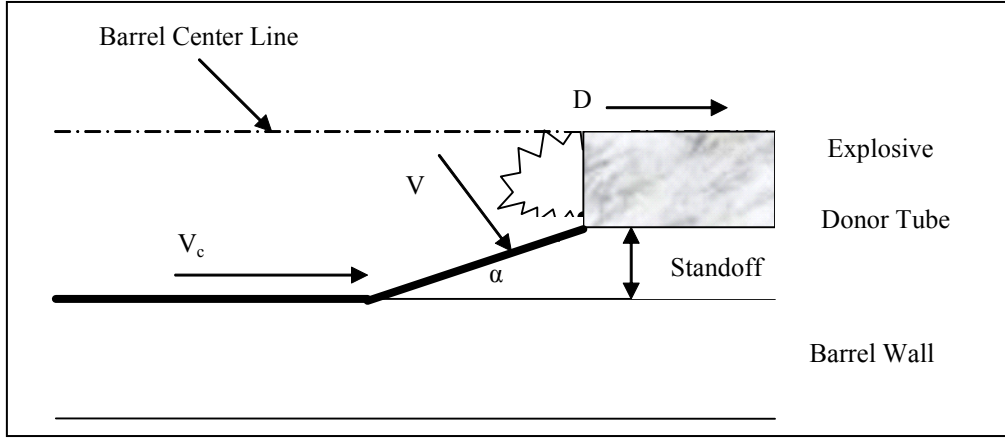


Figure 4. Geometry of explosive bonding setup.

In general,

$$V_c = D, \quad (4)$$

and

$$V = 2D \sin \alpha / 2. \quad (5)$$

For small values of  $\alpha$ ,

$$V = D \sin \alpha. \quad (6)$$

The first consideration is the minimum impact pressure needed to make the explosive bonding process work. Carpenter and Wittman (1975) provide an empirically-successful relation between the minimum donor tube impact velocity  $V_{\min}$  and the ultimate tensile strength  $\sigma_{ts}$  as follows:

$$V_{\min} = (\sigma_{ts} / \rho)^{1/2}, \quad (7)$$

where  $\rho$  is the donor tube density. It is presumed that the ultimate tensile strength is that strength measured at room temperature. However, it is expected that the donor tube will be heated during plastic deformation, lowering its strength. Consequently, the  $V_{\min}$  calculated may overestimate the actual value of  $V_{\min}$ . Note also that these same authors acknowledge that it would be better to use the Hugoniot elastic limit (HEL) to calculate  $V_{\min}$  (in another formula). However, the value of HEL for many alloys is not always available. Consequently, for sake of comparison among the alloys examined in this report, the ultimate tensile strength will be used in the calculations.

The second consideration is the existence of a specific collision velocity below, where researchers have found the bond line to be flat, and above, where they found it to be wavy. A wavy bond line is indicative of a good bond, implying that there is a lower limit to the collision velocity for a good bond. This transition velocity will be designated as  $V_T$ . Cowan et al. (1971) relate  $V_T$  to the density of the donor tube,  $\rho$  to the density of the gun barrel  $\rho_b$ , and  $H_F$  and  $H_B$  to



the diamond pyramid hardness of the donor tube and gun barrel, respectively, given a consistent set of units, in the following way:

$$V_T = \sqrt{2R_T(H_F + H_B)/(\rho + \rho_b)} . \quad (8)$$

$R_T$  is an empirically determined parameter that, for a wide range of metals, averages to 10.6 (no units). This is the value that will be used for calculations in this report.

Wittman (1973) derived a formula for the maximum donor tube velocity that would not result in melt-induced defects destroying the bond strength. This maximum velocity,  $V_{\max}$ , can be calculated from the following equation:

$$V_{\max} = \frac{(T_{MP} C_B)^{1/2} (KCC_B)^{1/4}}{N V_c (\rho h)^{1/4}} . \quad (9)$$

The material characteristics associated with the flyer plate are as follows:

- $T_{MP}$ , the melting point in °C;
- $C_B$ , the bulk sound speed;
- $K$ , the thermal conductivity;
- $C$ , the specific heat;
- $h$ , the flyer plate thickness; and
- $\rho$ , the flyer plate density.

$N$  is a constant that is not explicitly provided in the Carpenter and Wittman reference (1975). However, it can be derived from the table of material properties provided in this reference. First, calculate the value of  $V_{\min}$  using the values of  $\rho$  and  $\sigma_{ts}$  with equation 7. Next, determine  $NV_{\max}$  from the other parameters provided in Carpenter and Wittman (1975) and equation 9. The value of  $N$  can then be determined from the ratio of  $V_{\max}$  to  $V_{\min}$  provided in this reference. For the 12 metals listed, the average value of  $N$  is 0.11, with a mean deviation of 0.009.  $N$  will be taken as 0.11 (using cgs units) for calculations in this report.

There is experimental evidence that a jet is formed at the intersection of colliding surfaces during the explosive bonding process (Bergmann et al., 1966). It is generally accepted that this jet rids the colliding surfaces of any oxides and promotes a metallurgical bond. However, not all collisions result in a jet. Walsh et al. (1953) first proposed the concept of a critical collision angle for jet formation. This is the minimum angle at a specified collision velocity that is required for jet formation. Cowan et al. (1971) extended this work to asymmetric collisions. They give the angle  $\alpha$  in terms of the shock parameters and  $V_c$ :

$$\tan \alpha = U_p(V_c^2 - U_s^2)^{1/2} / (V_c^2 - U_p U_s) . \quad (10)$$

At the critical collision angle, the partial derivative of the pressure with respect to  $\alpha$  is zero (fixed  $V_c$ ) (Walsh et al., 1953). The pressure  $P$  is related to the shock velocity  $U_s$  and the particle velocity  $U_p$  through the following usual equation:

$$P = \rho U_s U_p. \quad (11)$$

The empirically determined relation between  $U_s$  and  $U_p$  is also required to determine the critical angle. The relation between the shock velocity and  $V_c$  is given by the following:

$$U_s = V_c \sin \beta, \quad (12)$$

where  $\beta$  is the angle between the shock front and the material flow vector into the collision point viewed from a frame of reference that is stationary with respect to the collision point. Rather than calculate  $\alpha$  explicitly in terms of  $P$  and take the partial derivative, it was easier to fix  $V_c$  and vary  $\beta$ . This generated values of  $U_s$ ,  $U_p$ ,  $P$ , and  $\alpha$ . A plot of  $P$  vs.  $\alpha$  showed a distinct maximum, and the critical angle for the given value of  $V_c$  was obtained. This was done for enough values of  $V_c$  to generate the required information.

## 4. Application of Governing Equations

Some cautions must be stated before applying the governing equations. First, they are to be applied with the understanding that the equations provide guidelines only. The actual parameters used to obtain the best possible bond will still be determined through a trial and error experimental process. Second, in applying these equations, it is important that a consistent set of units be used. In many instances, material property data are gathered from different sources that use different units. Some care must be exercised in converting all the data so that the units are consistent. Note that in equation 9, the value of  $N$  was determined using centimeter-gram-seconds °C as the set of units. Finally, it may not be possible to obtain the exact material properties for all alloys. In these cases, best estimates will be made.

While the use of pure tantalum resulted in a successful cladding of a liner to the M242 Bushmaster barrel, it may be that a higher-strength alloy is needed in a rifled bore configuration. Such alloys as tantalum-3% tungsten (Ta-3W) and Ta-10W are likely candidates. The densities of these two alloys can be found from a rule of mixtures, where the density of tantalum is taken to be  $16.65 \text{ g/cm}^3$  and the density of tungsten is taken to be  $19.3 \text{ g/cm}^3$ . The ultimate tensile strength of tantalum varies as a function of temperature. A value of  $250 \text{ MPa} = 36 \text{ ksi}$  is used for the room temperature value of  $\sigma_{ts}$  (American Society for Metals, 1979). The values of  $\sigma_{ts}$  for Ta-3W and Ta-10W are 60 ksi and 120 ksi, respectively (Aimone, 2004). The calculated values of  $V_{min}$  are given in table 1.

Table 1. Tantalum alloy material properties.

Material	Density (kg/m <sup>3</sup> )	Ultimate Tensile Strength, (MPa)	V <sub>min</sub> (km/s)	V <sub>T</sub> (km/s)
Tantalum	16.65 × 10 <sup>3</sup>	250	0.123	2.05
Ta-3W	16.73 × 10 <sup>3</sup>	414	0.157	2.10
Ta-10W	16.92 × 10 <sup>3</sup>	828	0.221	2.15

The transition velocity  $V_T$  for these materials can be found from equation 8. The Vickers hardness ranges for tantalum and Ta-3W are 90–100 and 110–130 (Aimone, 2004). An estimate of the hardness range for Ta-10W is 140–160 Vickers. Note that the hardness of the tantalum liner (after explosively bonding) was measured by Pepi et al. (2003) to be 140 Vickers. It is likely that the explosive bonding process increased the hardness of the liner above the original hardness. The gun steel hardness reported in Pepi et al. (2003) was 400 Vickers. For purposes here, the lower end of the hardness ranges of tantalum and its alloys before explosive bonding occurs will be used; the value of 400 Vickers will be used for the steel. The calculated values for  $V_T$  are also shown in table 1.

It is expected that the critical angle for jet formation and  $V_{max}$  will depend on the bulk properties of the material so that a calculation for tantalum will provide close approximations of these values for the Ta-3W and Ta-10W alloys (Furnish et al., 1995 ). The values of the bulk properties of tantalum used to calculate  $V_{max}$  are shown in table 2. The relation between the shock velocity  $U_s$  and the particle velocity  $U_p$  (Marsh, 1995) is given by the following:

$$U_s = 3.43 \text{ (km/s)} + 1.19 U_p. \quad (13)$$

In this equation,  $U_s$  and  $U_p$  are in kilometers per second.

Table 2. Tantalum material properties.

Property	Value	Reference
Bulk sound speed	3.43 km/s	(Marsh, 1980)
Density	16.65 × 10 <sup>3</sup> kg/m <sup>3</sup>	(American Society for Metals, 1979)
Melting point	3269° K	(American Society for Metals, 1979)
Thermal conductivity	54.4 w/m K	(American Society for Metals, 1979)
Specific heat	139.1 J/kg K	(American Society for Metals, 1979)

A flyer plate thickness of 0.15 cm (0.059 in) was selected. The table 2 values of input parameters were converted to the cgs system, and equation 9 was used to find  $V_{max}$  in terms of  $V_c$ .

$$V_{max} = 0.923 \times 10^{10} / V_c, \quad (14)$$

where  $V_{max}$  and  $V_c$  are now in centimeters per second. We then have the following condition on  $\alpha$  at  $V = V_{max}$ :

$$\alpha = 2\sin^{-1}(V_{max}/2V_c) = 2\sin^{-1}(.923/2V_c^2), \quad (15)$$

with  $V_{\max}$  and  $V_c$  in kilometers per second.

The calculation for the critical angle for jet formation is as follows. First, fix a value of  $V_c$ . Then,

$$U_s = V_c \sin\beta. \quad (16)$$

From the relation between  $U_s$  and  $U_p$ ,

$$U_p = (V_c \sin\beta - 3.43)/1.19. \quad (17)$$

Also,

$$P = \rho (V_c \sin\beta)(V_c \sin\beta - 3.43)/1.19. \quad (18)$$

Then, from equation 10,

$$\tan \alpha = ((\sin\beta - 3.43/V_c)/1.19)(1 - \sin^2\beta)^{1/2}/(1 - \sin\beta(\sin\beta - 3.43/V_c)/1.19). \quad (19)$$

By varying  $\beta$ , values of  $P$  and  $\alpha$  can be generated. A plot of  $P$  vs.  $\alpha$  for the case of  $V_c = 5$  is shown in figure 5.

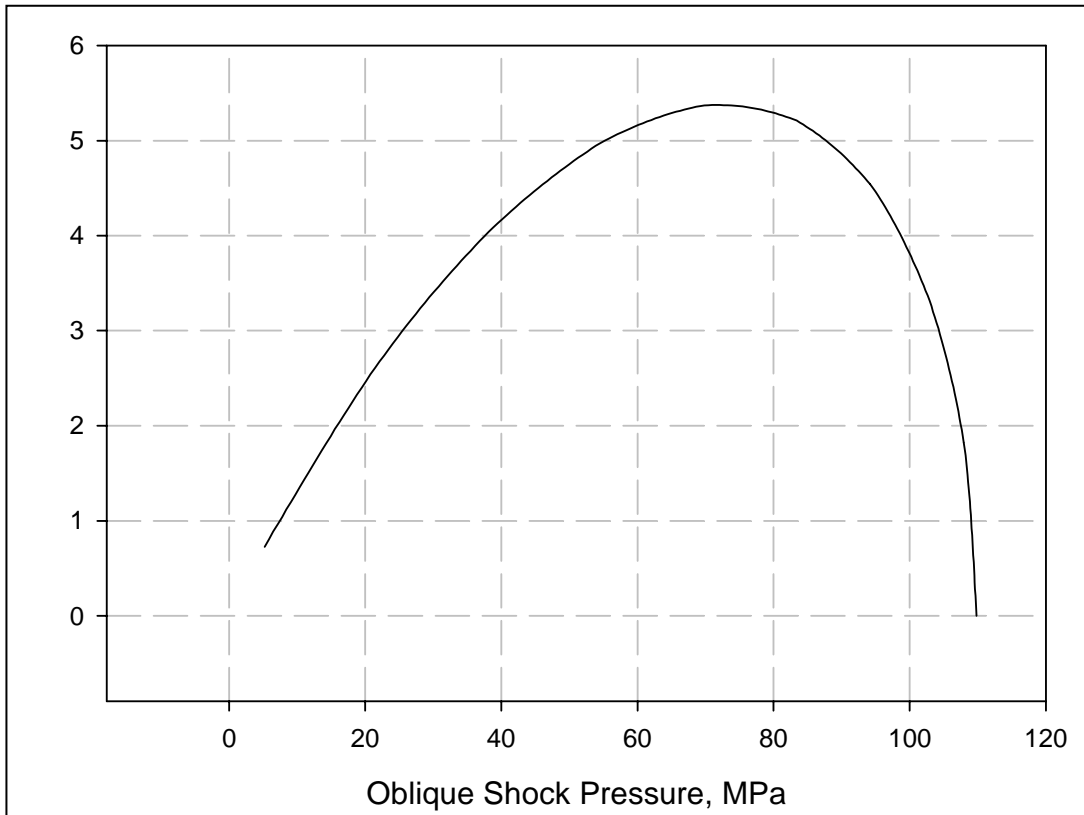


Figure 5. Relation of pressure  $P$  in MPa vs.  $\alpha$  in degrees for  $V_c = 5$  km/s.

It is clear from this figure that the maximum value of  $\alpha$  is  $\sim 5.4^\circ$ . Thus, the critical angle for  $V_c = 5 \text{ km/s}$  is  $5.4^\circ$ . The relation between  $V_c$  and  $\alpha$  can be generated with this approach.

The end result of all these calculations is a plot similar to that shown in Carpenter and Wittman (1975), shown in figure 6. For each metal, there is a central area bounded by four lines that indicate appropriate ranges of parameters. Bounded from below are the plots of  $V_{\min}$  for each of the metals. On the left, the boundary is  $V_T$ . The values for each metal are so close together that  $V_T$  is represented by a single thick vertical line. On the right, the area is bounded by the condition for the critical angle for collision. At the top, the area is bounded by  $V_{\max}$ .

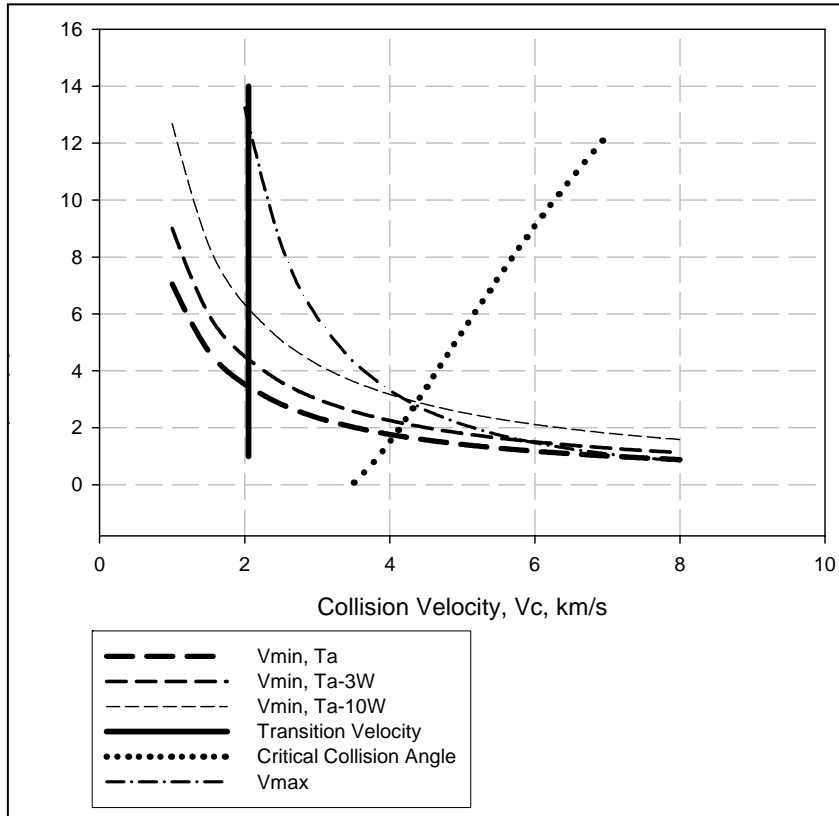


Figure 6. Bounding plots for explosive bonding of tantalum and two alloys.

A similar plot can be made for niobium (annealed). Table 3 lists the properties that were used for the calculations. All values were obtained from the Metals Handbook (American Society for Metals, 1979) except for the bulk sound speed, obtained from Marsh (1980). In addition, the relation

$$U_s = 4.46 \text{ (km/s)} + 1.20U_p \quad (20)$$

and a value of 0.15 cm for the flyer plate thickness were used to generate the plots in figure 7.

Considering the fact that the material properties for niobium are not too different from those of tantalum and its alloys, it is not surprising that the bounding plots in figure 7 appear similar to those in figure 6.

Table 3. Properties of niobium.

Property	Value
Tensile strength	275 MPa
Density	8.57 g/cm <sup>3</sup>
Hardness	80 Vickers
Melting point	2468 °C
Bulk sound speed	4.46 km/s
Specific heat (at 20° C)	270 J/kg K
Thermal conductivity (at 0° C)	52.3 w/m K

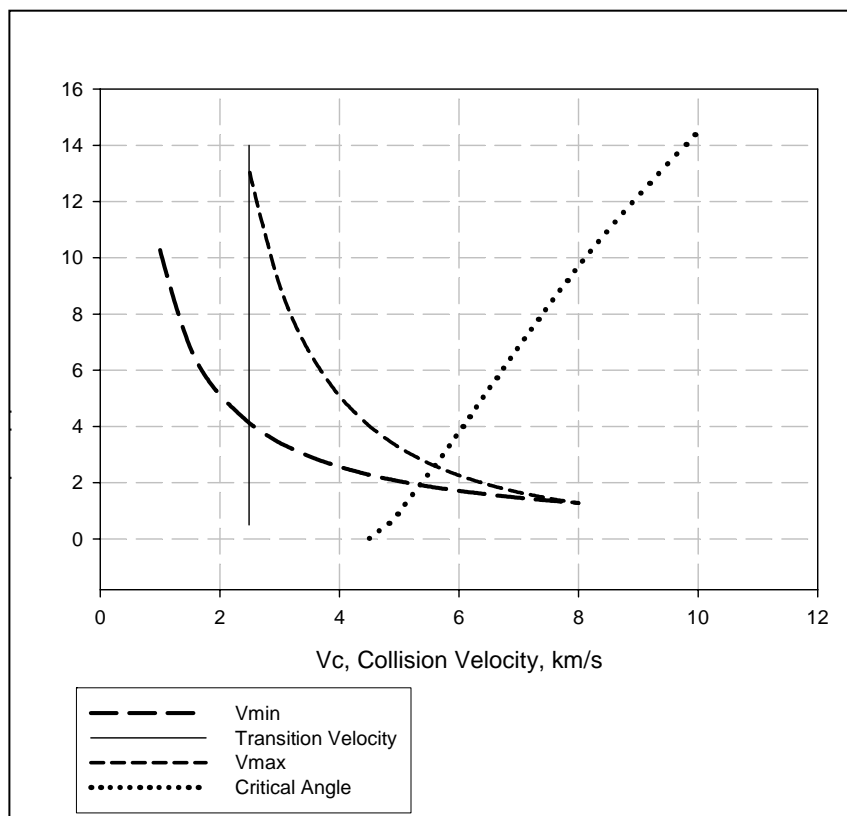


Figure 7. Bounding plots for niobium.

## 5. Discussion

The governing equations presented in section 3 have been developed as a result of many years of experience in explosive welding. While they are based on a physical understanding of the explosive bonding process, reliance is also placed on fixing certain parameters obtained from averages of a large number of tests. For this reason, the governing equations should be taken as guidelines to establish the approximate operating parameters for a specific case. Note also that the equations were based primarily on experience gained with cladding flat pats on a flat

substrate. In the case of explosively bonding a liner to the inside of a gun tube, the geometry is different. The large amount of hoop strain that might occur in bonding liners could affect material properties to the extent that the equations are no longer good approximations. This is especially true if the material properties are highly strain and/or strain rate sensitive.

In any event, the governing equations help to assess the cladability of a material. Perhaps the most important material property is the ultimate tensile strength. As seen in figure 6, as the ultimate tensile strength of the various tantalum alloys increases, the  $V_c$ - $\alpha$  operating area decreases. This is because the lower limit ( $V_{min}$ ) is raised, but the upper limit, being based on bulk properties such as density and sound speed, stays about the same. Consequently, the choice of high explosive is more restrictive as small alloy additions increase a material's strength. Note that if annealing can lower the tensile strength of the Ta-10W, its cladability would improve (Montgomery, 2004).

The choice of explosive will have a great deal to do with the cladability of a material. The explosive used by TPL had an upper limit on the detonation velocity  $D$  of 2.2 km/s. Since  $D = V_c$ , TPL may have been operating at the far left portion of the bounded area for tantalum and the two alloys.  $V_T$  for Ta-10W was calculated to be 2.15 km/s, providing another possibility as to why efforts to form a good bond were not successful with this material. For  $V = 0.212$  km/s and  $D = 2.2$  km/s,  $\alpha = 5.5^\circ$ . This value appears to lie close to the  $V_{min}$  line shown in figure 6 and may provide another reason why the use of this particular explosive was not successful in bonding the Ta-10W. However, the explosive should work for the Ta-3W.

The value of  $\alpha$  can be varied by changing  $V$  where  $D$  is fixed (equations 5 or 6). This can be done by changing the charge-to-mass ratio (see equation 2). The possible range of  $\alpha$  can be calculated with the following assumptions. First, we expect that the liner will be fully packed with explosive since allowing a hollow portion down the axis of the explosive may result in a dimension less than the failure diameter. Next, consider standoffs between 0.01 and 0.25 in. This will allow the mass of both the metal liner and explosive to vary. Finally, we use the dimensions previously discussed (1.064-in inner diameter of the gun tube and a final liner thickness of 0.044 in).

The relation between  $\alpha$  and standoff is shown in figure 8. The larger standoffs reduce the amount of explosive and increase the thickness of the metal liner. Thus, the value of  $V$  will decrease for larger standoffs, resulting in a decrease in  $\alpha$ . Referring to figure 6, it can be seen that the values of  $\alpha$  in the range of 4–8° at  $V_c = 2.2$  km/s are acceptable and may even allow explosive bonding of the higher-strength alloys for the higher values of  $\alpha$  (shorter standoff).

A final consideration in choice of explosives is failure diameter. For all explosives, there is a minimum dimension needed to sustain a detonation without confinement. In general, explosives with low detonation velocities such as ammonium nitrate have large failure diameters. For the current application, this minimum dimension must be less than 25 mm, the bore diameter of the M242 gun tube. This condition limits the available choices of explosives for bonding the liner to the gun tube.

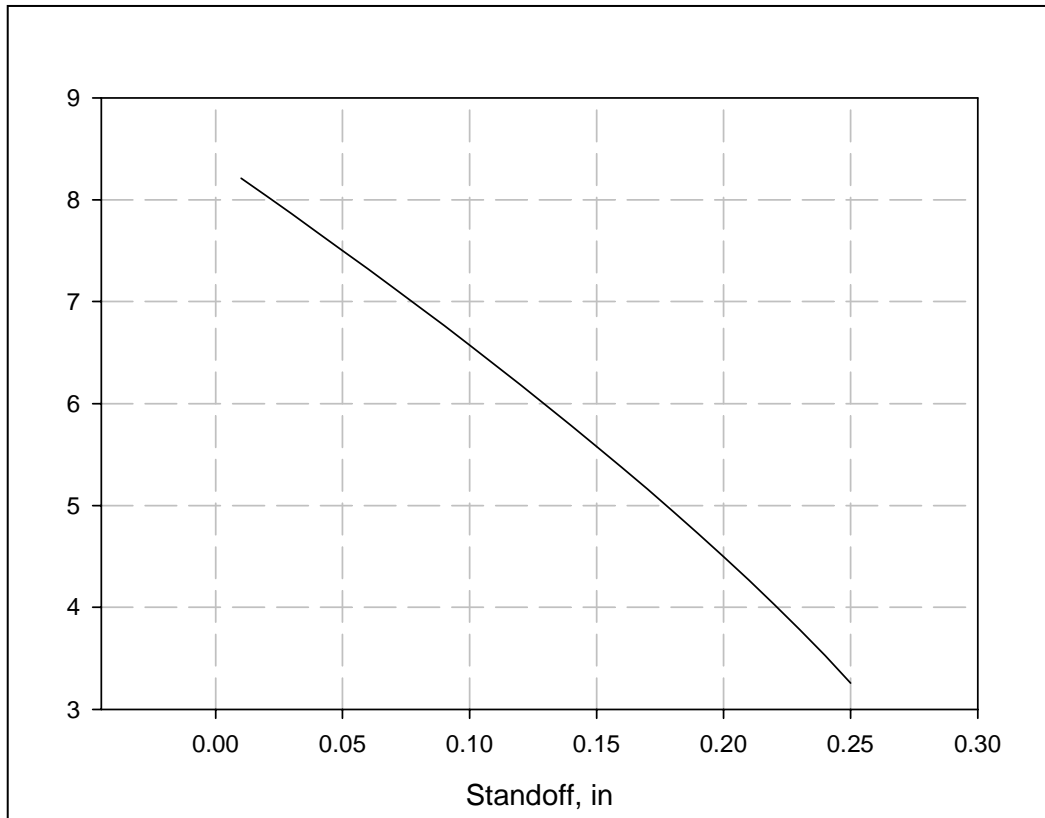


Figure 8. Relation between  $\alpha$  (in degrees) and standoff for a specific case.

---

## 6. Summary

---

A pure tantalum liner has been successfully bonded to the inside of an M242 Bushmaster medium-caliber cannon in previous work performed by TPL, Inc. under a Phase 2 SBIR. In expectation that the tantalum will be too soft to resist wear forces at the lands and grooves in the gun barrel, other alloys were examined for suitability for cladding. Semiempirical equations governing explosive bonding were applied to two tantalum alloys and niobium. The equations indicated that for the particular explosive used by TPL, a shorter standoff would facilitate the bonding of higher-strength alloys by increasing the collision angle  $\alpha$  and reducing the amount of strain in the liner produced by the explosive bonding process.



---

## 7. References

---

- Aimone, Paul. H. C. Starck Co. Private communication, March 2004.
- American Society for Metals. *Metals Handbook, 9th Edition, Volume 2—Properties and Selection: Nonferrous Alloys and Pure Metals*; Metals Park, OH, 1979.
- Bergmann, O. R.; Cowan, G. R.; Holtzman, A. H. Experimental Evidence of Jet Formation During Explosion Cladding. *Transactions of the Metallurgical Society of AIME* **1966**, 236, 646.
- Carpenter, S. H.; Wittman, R. H. Explosion Welding. *Ann. Rev. Mat. Sci.* **1975**, 5.
- Cooper, P. W. *Basics of Explosive Engineering*; VCH: New York, 1996.
- Cowan, G. R., Bergman, O. R.; Holtzman, A. H. Mechanism of Bond Zone Wave Formation in Explosion-Clad Metals. *Met. Trans.* **1971**, 2.
- Furnish, M. D.; Lassila, D. H.; Chhabildas, L. C.; Steinberg, D. J. *Dynamic Material Properties of Refractory Metals: Tantalum and Tantalum/Tungsten Alloys, in Shock Compression of Matter—1995*; Schmidt, S. C., Tao, W. C., Eds.; AIP Press: Woodbury, NY, 1995.
- Lowey, R. F. *Gun Tube Liner Erosion and Wear Protection*; TPL-FR-ER31; under contract DAAD19-99-C-0002, TPL, Inc.: Albuquerque, NM, 28 May 2002.
- Lowey, R. F. TPL, Inc., Albuquerque, NM. Private communication, 2004a.
- Lowey, R. F. TPL, Inc., Albuquerque, NM. Private communication, 2004b.
- Marsh, S. P., Ed. *LASL Shock Hugoniot Data*; University of California Press: Berkley, CA, 1980.
- Montgomery, J. U.S. Army Research Laboratory, Aberdeen Proving Ground, MD. Private communication, April 2004.
- Mulligan, C.; Audino, M.; Cote, P.; Kendall, G.; Rickard, C.; Smith, S.; Todaro, M. *Characterization of Explosively Bonded and Fired Tantalum Liners Applied to 25-mm Gun Tubes*; ARCCB-TR-02016; Benet Laboratories: Watervliet, NY, November 2002.
- Pepi, M.; Snoha, D. J.; Montgomery, J. S.; de Rosset, W. S. *Examination of Intermetallic Phases and Residual Stresses Resulting from Explosive Bonding of Refractory Metal Gun Tube Liners*; ARL-MR-550; U.S. Army Research Laboratory: Aberdeen Proving Ground, MD, February 2003.

- Walsh, J. M.; Schreffler, R. G.; Willig, F. J. Limiting Conditions for Jet Formation in High Velocity Collisions. *Journal of Applied Physics* **1953**, 24 (3).
- Wittman, R. H. *Proceedings of the Second International Symposium Use Explosive Energy Mfg. Metallic Materials*, Marianske-Lazne, Czechoslovakia, 1973.
- Wylie, H. K.; Williams, P. E. G.; Crossland, B. An Experimental Investigation of Explosive Welding Parameters. *Proceedings of the First International Symposium, Use of Explosive Energy in Manufacturing Metallic Materials of New Properties and Possibilities of Application Thereof in Chemical Industry*, Marianske-Lazne, Czechoslovakia, 1970.
- Zukas, J. A.; Walters, W. P. *Explosive Effects and Applications*; Springer-Verlag: New York, NY, 1998.

NO. OF  
COPIES ORGANIZATION

1  
(PDF  
Only) DEFENSE TECHNICAL  
INFORMATION CTR  
DTIC OCA  
8725 JOHN J KINGMAN RD  
STE 0944  
FT BELVOIR VA 22060-6218

1 COMMANDING GENERAL  
US ARMY MATERIEL CMD  
AMCRDA TF  
5001 EISENHOWER AVE  
ALEXANDRIA VA 22333-0001

1 INST FOR ADVNCD TCHNLGY  
THE UNIV OF TEXAS  
AT AUSTIN  
3925 W BRAKER LN STE 400  
AUSTIN TX 78759-5316

1 US MILITARY ACADEMY  
MATH SCI CTR EXCELLENCE  
MADN MATH  
THAYER HALL  
WEST POINT NY 10996-1786

1 DIRECTOR  
US ARMY RESEARCH LAB  
AMSRD ARL CS IS R  
2800 POWDER MILL RD  
ADELPHI MD 20783-1197

3 DIRECTOR  
US ARMY RESEARCH LAB  
AMSRD ARL CI OK TL  
2800 POWDER MILL RD  
ADELPHI MD 20783-1197

3 DIRECTOR  
US ARMY RESEARCH LAB  
AMSRD ARL CS IS T  
2800 POWDER MILL RD  
ADELPHI MD 20783-1197

NO. OF  
COPIES ORGANIZATION

ABERDEEN PROVING GROUND

1 DIR USARL  
AMSRD ARL CI OK TP (BLDG 4600)

NO. OF  
COPIES ORGANIZATION

1 DIRECTOR  
US ARMY RESEARCH LAB  
AMSRD ARL SE L  
D SNIDER  
2800 POWDER MILL RD  
ADELPHI MD 20783-1197

1 DIRECTOR  
US ARMY RESEARCH LAB  
AMSRD ARL SE DE  
R ATKINSON  
2800 POWDER MILL RD  
ADELPHI MD 20783-1197

5 DIRECTOR  
US ARMY RESEARCH LAB  
AMSRD ARL WM MB  
A ABRAHAMIAN  
M BERMAN  
M CHOWDHURY  
T LI  
E SZYMANSKI  
2800 POWDER MILL RD  
ADELPHI MD 20783-1197

1 COMMANDER  
US ARMY MATERIEL CMD  
AMXMI INT  
5001 EISENHOWER AVE  
ALEXANDRIA VA 22333-0001

2 PM MAS  
SFAE AMO MAS MC  
PICATINNY ARSENAL NJ  
07806-5000

3 COMMANDER  
US ARMY ARDEC  
AMSTA AR CC  
M PADGETT  
J HEDDERICH  
H OPAT  
PICATINNY ARSENAL NJ  
07806-5000

2 COMMANDER  
US ARMY ARDEC  
AMSTA AR AE WW  
E BAKER  
J PEARSON  
PICATINNY ARSENAL NJ  
07806-5000

NO. OF  
COPIES ORGANIZATION

1 COMMANDER  
US ARMY ARDEC  
AMSTA AR FSE  
PICATINNY ARSENAL NJ  
07806-5000

1 COMMANDER  
US ARMY ARDEC  
AMSTA AR TD  
PICATINNY ARSENAL NJ  
07806-5000

13 COMMANDER  
US ARMY ARDEC  
AMSTA AR CCH A  
F ALTAMURA  
M NICOLICH  
M PALATHINGUL  
D VO  
R HOWELL  
A VELLA  
M YOUNG  
L MANOLE  
S MUSALLI  
R CARR  
M LUCIANO  
E LOGSDEN  
T LOUZEIRO  
PICATINNY ARSENAL NJ  
07806-5000

1 COMMANDER  
US ARMY ARDEC  
AMSTA AR CCH P  
J LUTZ  
PICATINNY ARSENAL NJ  
07806-5000

1 COMMANDER  
US ARMY ARDEC  
AMSTA AR FSF T  
C LIVECCHIA  
PICATINNY ARSENAL NJ  
07806-5000

1 COMMANDER  
US ARMY ARDEC  
AMSTA ASF  
PICATINNY ARSENAL NJ  
07806-5000

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	COMMANDER US ARMY ARDEC AMSTA AR QAC T C J PAGE PICATINNY ARSENAL NJ 07806-5000
1	COMMANDER US ARMY ARDEC AMSTA AR M D DEMELLA PICATINNY ARSENAL NJ 07806-5000
3	COMMANDER US ARMY ARDEC AMSTA AR FSA A WARNASH B MACHAK M CHIEFA PICATINNY ARSENAL NJ 07806-5000
2	COMMANDER US ARMY ARDEC AMSTA AR FSP G M SCHIKSNIS D CARLUCCI PICATINNY ARSENAL NJ 07806-5000
2	COMMANDER US ARMY ARDEC AMSTA AR CCH C H CHANIN S CHICO PICATINNY ARSENAL NJ 07806-5000
1	COMMANDER US ARMY ARDEC AMSTA AR QAC T D RIGOGLIOSO PICATINNY ARSENAL NJ 07806-5000
1	COMMANDER US ARMY ARDEC AMSTA AR WET T SACHAR BLDG 172 PICATINNY ARSENAL NJ 07806-5000

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	US ARMY ARDEC INTELLIGENCE SPECIALIST AMSTA AR WEL F M GUERRIERE PICATINNY ARSENAL NJ 07806-5000
10	COMMANDER US ARMY ARDEC AMSTA AR CCH B P DONADIA F DONLON P VALENTI C KNUTSON G EUSTICE K HENRY J MCNABOC G WAGNECZ R SAYER F CHANG PICATINNY ARSENAL NJ 07806-5000
6	COMMANDER US ARMY ARDEC AMSTA AR CCL F PUZYCKI R MCHUGH D CONWAY E JAROSZEWSKI R SCHLENNER M CLUNE PICATINNY ARSENAL NJ 07806-5000
1	PM ARMS SFAE GCSS ARMS BLDG 171 PICATINNY ARSENAL NJ 07806-5000
1	COMMANDER US ARMY ARDEC AMSTA AR WEA J BRESCIA PICATINNY ARSENAL NJ 07806-5000
1	PM MAS SFAE AMO MAS PICATINNY ARSENAL NJ 07806-5000

NO. OF  
COPIES ORGANIZATION

1 PM MAS  
SFAE AMO MAS  
CHIEF ENGINEER  
PICATINNY ARSENAL NJ  
07806-5000

1 PM MAS  
SFAE AMO MAS PS  
PICATINNY ARSENAL NJ  
07806-5000

2 PM MAS  
SFAE AMO MAS LC  
PICATINNY ARSENAL NJ  
07806-5000

1 COMMANDER  
US ARMY ARDEC  
PRODUCTION BASE  
MODERN ACTY  
AMSMC PBM K  
PICATINNY ARSENAL NJ  
07806-5000

1 COMMANDER  
US ARMY TACOM  
PM COMBAT SYSTEMS  
SFAE GCS CS  
6501 ELEVEN MILE RD  
WARREN MI 48397-5000

1 COMMANDER  
US ARMY TACOM  
AMSTA SF  
WARREN MI 48397-5000

1 DIRECTOR  
AIR FORCE RESEARCH LAB  
MLLMD  
D MIRACLE  
2230 TENTH ST  
WRIGHT PATTERSON AFB OH  
45433-7817

1 OFC OF NAVAL RESEARCH  
J CHRISTODOULOU  
ONR CODE 332  
800 N QUINCY ST  
ARLINGTON VA 22217-5600

1 US ARMY CERL  
R LAMPO  
2902 NEWMARK DR  
CHAMPAIGN IL 61822

NO. OF  
COPIES ORGANIZATION

1 COMMANDER  
US ARMY TACOM  
PM SURVIVABLE SYSTEMS  
SFAE GCSS W GSI H  
M RYZYI  
6501 ELEVEN MILE RD  
WARREN MI 48397-5000

1 COMMANDER  
US ARMY TACOM  
CHIEF ABRAMS TESTING  
SFAE GCSS W AB QT  
T KRASKIEWICZ  
6501 ELEVEN MILE RD  
WARREN MI 48397-5000

1 COMMANDER  
WATERVLIET ARSENAL  
SMCWV QAE Q  
B VANINA  
BLDG 44  
WATERVLIET NY 12189-4050

1 TNG, DOC, & CBT DEV  
ATZK TDD IRSA  
A POMEY  
FT KNOX KY 40121

2 HQ IOC TANK  
AMMUNITION TEAM  
AMSIO SMT  
R CRAWFORD  
W HARRIS  
ROCK ISLAND IL 61299-6000

2 COMMANDER  
US ARMY AMCOM  
AVIATION APPLIED TECH DIR  
J SCHUCK  
FT EUSTIS VA 23604-5577

1 NSWC  
DAHLGREN DIV CODE G06  
DAHLGREN VA 22448

2 US ARMY CORPS OF ENGR  
CERD C  
T LIU  
CEW ET  
T TAN  
20 MASSACHUSETTS AVE NW  
WASHINGTON DC 20314

NO. OF  
COPIES ORGANIZATION

1 US ARMY COLD REGIONS  
RSCH & ENGRNG LAB  
P DUTTA  
72 LYME RD  
HANOVER NH 03755

14 COMMANDER  
US ARMY TACOM  
AMSTA TR R  
R MCCLELLAND  
D THOMAS  
J BENNETT  
D HANSEN  
AMSTA JSK  
S GOODMAN  
J FLORENCE  
K IYER  
D TEMPLETON  
A SCHUMACHER  
AMSTA TR D  
D OSTBERG  
L HINOJOSA  
B RAJU  
AMSTA CS SF  
H HUTCHINSON  
F SCHWARZ  
WARREN MI 48397-5000

14 BENET LABS  
AMSTA AR CCB  
R FISCELLA  
M SOJA  
E KATHE  
M SCAVULO  
G SPENCER  
P WHEELER  
S KRUPSKI  
J VASILAKIS  
G FRIAR  
R HASENBEIN  
AMSTA CCB R  
S SOPOK  
E HYLAND  
D CRAYON  
R DILLON  
WATERVLIET NY 12189-4050

1 USA SBCCOM PM SOLDIER SPT  
AMSSB PM RSS A  
J CONNORS  
KANSAS ST  
NATICK MA 01760-5057

NO. OF  
COPIES ORGANIZATION

1 NSW  
TECH LIBRARY CODE 323  
17320 DAHLGREN RD  
DAHLGREN VA 22448

2 USA SBCCOM  
MATERIAL SCIENCE TEAM  
AMSSB RSS  
J HERBERT  
M SENNETT  
KANSAS ST  
NATICK MA 01760-5057

2 OFC OF NAVAL RESEARCH  
D SIEGEL CODE 351  
J KELLY  
800 N QUINCY ST  
ARLINGTON VA 22217-5660

1 NSW  
CRANE DIVISION  
M JOHNSON CODE 20H4  
LOUISVILLE KY 40214-5245

2 NSW  
U SORATHIA  
C WILLIAMS CD 6551  
9500 MACARTHUR BLVD  
WEST BETHESDA MD 20817

2 COMMANDER  
NSWC  
CARDEROCK DIVISION  
R PETERSON CODE 2020  
M CRITCHFIELD CODE 1730  
BETHESDA MD 20084

8 DIRECTOR  
US ARMY NGIC  
D LEITER MS 404  
M HOLTUS MS 301  
M WOLFE MS 307  
S MINGLEDORF MS 504  
J GASTON MS 301  
W GSTATTENBAUER MS 304  
R WARNER MS 305  
J CRIDER MS 306  
2055 BOULDERS RD  
CHARLOTTESVILLE VA  
22911-8318

NO. OF COPIES	ORGANIZATION
1	NAVAL SEA SYSTEMS CMD D LIESE 1333 ISAAC HULL AVE SE 1100 WASHINGTON DC 20376-1100
1	EXPEDITIONARY WARFARE DIV N85 F SHOUP 2000 NAVY PENTAGON WASHINGTON DC 20350-2000
8	US ARMY SBCCOM SOLDIER SYSTEMS CENTER BALLISTICS TEAM J WARD W ZUKAS P CUNNIFF J SONG MARINE CORPS TEAM J MACKIEWICZ BUS AREA ADVOCACY TEAM W HASKELL AMSSB RCP SS W NYKVIST S BEAUDOIN KANSAS ST NATICK MA 01760-5019
7	US ARMY RESEARCH OFC A CROWSON H EVERETT J PRATER G ANDERSON D STEPP D KISEROW J CHANG PO BOX 12211 RESEARCH TRIANGLE PARK NC 27709-2211
1	AFRL MLBC 2941 P ST RM 136 WRIGHT PATTERSON AFB OH 45433-7750
1	DIRECTOR LOS ALAMOS NATL LAB F L ADDESSIO T 3 MS 5000 PO BOX 1633 LOS ALAMOS NM 87545

NO. OF COPIES	ORGANIZATION
8	NSWC J FRANCIS CODE G30 D WILSON CODE G32 R D COOPER CODE G32 J FRAYSSE CODE G33 E ROWE CODE G33 T DURAN CODE G33 L DE SIMONE CODE G33 R HUBBARD CODE G33 DAHLGREN VA 22448
1	NSWC CARDEROCK DIVISION R CRANE CODE 6553 9500 MACARTHUR BLVD WEST BETHESDA MD 20817-5700
1	AFRL MLSS R THOMSON 2179 12TH ST RM 122 WRIGHT PATTERSON AFB OH 45433-7718
2	AFRL F ABRAMS J BROWN BLDG 653 2977 P ST STE 6 WRIGHT PATTERSON AFB OH 45433-7739
5	DIRECTOR LLNL R CHRISTENSEN S DETERESA F MAGNESS M FINGER MS 313 M MURPHY L 282 PO BOX 808 LIVERMORE CA 94550
1	AFRL MLS OL L COULTER 5851 F AVE BLDG 849 RM AD1A HILL AFB UT 84056-5713
1	OSD JOINT CCD TEST FORCE OSD JCCD R WILLIAMS 3909 HALLS FERRY RD VICKSBURG MS 29180-6199



NO. OF  
COPIES ORGANIZATION

3 DARPA  
M VANFOSSSEN  
S WAX  
L CHRISTODOULOU  
3701 N FAIRFAX DR  
ARLINGTON VA 22203-1714

2 SERDP PROGRAM OFC  
PM P2  
C PELLERIN  
B SMITH  
901 N STUART ST STE 303  
ARLINGTON VA 22203

1 OAK RIDGE NATL LAB  
R M DAVIS  
PO BOX 2008  
OAK RIDGE TN 37831-6195

1 OAK RIDGE NATL LAB  
C EBERLE MS 8048  
PO BOX 2008  
OAK RIDGE TN 37831

3 DIRECTOR  
SANDIA NATL LABS  
APPLIED MECHS DEPT  
MS 9042  
J HANDROCK  
Y R KAN  
J LAUFFER  
PO BOX 969  
LIVERMORE CA 94551-0969

1 OAK RIDGE NATL LAB  
C D WARREN MS 8039  
PO BOX 2008  
OAK RIDGE TN 37831

4 NIST  
M VANLANDINGHAM MS 8621  
J CHIN MS 8621  
J MARTIN MS 8621  
D DUTHINH MS 8611  
100 BUREAU DR  
GAITHERSBURG MD 20899

1 HYDROGEOLOGIC INC  
SERDP ESTCP SPT OFC  
S WALSH  
1155 HERNDON PKWY STE 900  
HERNDON VA 20170

NO. OF  
COPIES ORGANIZATION

3 NASA LANGLEY RESEARCH CTR  
AMSRD ARL VS  
W ELBER MS 266  
F BARTLETT JR MS 266  
G FARLEY MS 266  
HAMPTON VA 23681-0001

1 NASA LANGLEY RESEARCH CTR  
T GATES MS 188E  
HAMPTON VA 23661-3400

1 FHWA  
E MUNLEY  
6300 GEORGETOWN PIKE  
MCLEAN VA 22101

1 USDOT FEDERAL RAILROAD  
M FATEH RDV 31  
WASHINGTON DC 20590

3 CYTEC FIBERITE  
R DUNNE  
D KOHLI  
R MAYHEW  
1300 REVOLUTION ST  
HAVRE DE GRACE MD 21078

1 DIRECTOR  
NGIC  
IANG TMT  
2055 BOULDERS RD  
CHARLOTTESVILLE VA  
22911-8318

1 SIOUX MFG  
B KRIEL  
PO BOX 400  
FT TOTTEN ND 58335

2 3TEX CORP  
A BOGDANOVICH  
J SINGLETARY  
109 MACKENAN DR  
CARY NC 27511

1 3M CORP  
J SKILDUM  
3M CENTER BLDG 60 IN 01  
ST PAUL MN 55144-1000

NO. OF  
COPIES ORGANIZATION

1 DIRECTOR  
DEFENSE INTLLGNC AGENCY  
TA 5  
K CRELLING  
WASHINGTON DC 20310

1 ADVANCED GLASS FIBER YARNS  
T COLLINS  
281 SPRING RUN LANE STE A  
DOWNINGTON PA 19335

1 COMPOSITE MATERIALS INC  
D SHORTT  
19105 63 AVE NE  
PO BOX 25  
ARLINGTON WA 98223

1 JPS GLASS  
L CARTER  
PO BOX 260  
SLATER RD  
SLATER SC 29683

1 COMPOSITE MATERIALS INC  
R HOLLAND  
11 JEWEL CT  
ORINDA CA 94563

1 COMPOSITE MATERIALS INC  
C RILEY  
14530 S ANSON AVE  
SANTA FE SPRINGS CA 90670

2 SIMULA  
J COLTMAN  
R HUYETT  
10016 S 51ST ST  
PHOENIX AZ 85044

2 PROTECTION MATERIALS INC  
M MILLER  
F CRILLEY  
14000 NW 58 CT  
MIAMI LAKES FL 33014

2 FOSTER MILLER  
M ROYLANCE  
W ZUKAS  
195 BEAR HILL RD  
WALTHAM MA 02354-1196

NO. OF  
COPIES ORGANIZATION

1 ROM DEVELOPMENT CORP  
R O MEARA  
136 SWINEBURNE ROW  
BRICK MARKET PLACE  
NEWPORT RI 02840

2 TEXTRON SYSTEMS  
T FOLTZ  
M TREASURE  
1449 MIDDLESEX ST  
LOWELL MA 01851

1 O GARA HESS & EISENHARDT  
M GILLESPIE  
9113 LESAINTE DR  
FAIRFIELD OH 45014

2 MILLIKEN RESEARCH CORP  
H KUHN  
M MACLEOD  
PO BOX 1926  
SPARTANBURG SC 29303

1 CONNEAUGHT INDUSTRIES INC  
J SANTOS  
PO BOX 1425  
COVENTRY RI 02816

1 ARMTEC DEFENSE PRODUCTS  
S DYER  
85 901 AVE 53  
PO BOX 848  
COACHELLA CA 92236

1 NATL COMPOSITE CTR  
T CORDELL  
2000 COMPOSITE DR  
KETTERING OH 45420

3 PACIFIC NORTHWEST LAB  
M SMITH  
G VAN ARSDALE  
R SHIPPELL  
PO BOX 999  
RICHLAND WA 99352

1 SAIC  
M PALMER  
1410 SPRING HILL RD STE 400  
MS SH4 5  
MCLEAN VA 22102

NO. OF  
COPIES ORGANIZATION

1 ALLIANT TECHSYSTEMS INC  
4700 NATHAN LN N  
PLYMOUTH MN 55442-2512

1 APPLIED COMPOSITES  
W GRISCH  
333 NORTH SIXTH ST  
ST CHARLES IL 60174

1 CUSTOM ANALYTICAL  
ENG SYS INC  
A ALEXANDER  
13000 TENSOR LANE NE  
FLINTSTONE MD 21530

1 AAI CORP  
DR N B MCNELLIS  
PO BOX 126  
HUNT VALLEY MD 21030-0126

1 OFC DEPUTY UNDER SEC DEFNS  
J THOMPSON  
1745 JEFFERSON DAVIS HWY  
CRYSTAL SQ 4 STE 501  
ARLINGTON VA 22202

3 ALLIANT TECHSYSTEMS INC  
J CONDON  
E LYNAM  
J GERHARD  
WV01 16 STATE RT 956  
PO BOX 210  
ROCKET CENTER WV  
26726-0210

1 PROJECTILE TECHNOLOGY INC  
515 GILES ST  
HAVRE DE GRACE MD 21078

1 HEXCEL INC  
R BOE  
PO BOX 18748  
SALT LAKE CITY UT 84118

1 PRATT & WHITNEY  
C WATSON  
400 MAIN ST MS 114 37  
EAST HARTFORD CT 06108

NO. OF  
COPIES ORGANIZATION

5 NORTHROP GRUMMAN  
B IRWIN  
K EVANS  
D EWART  
A SHREKENHAMER  
J MCGLYNN  
BLDG 160 DEPT 3700  
1100 WEST HOLLYVALE ST  
AZUSA CA 91701

1 HERCULES INC  
HERCULES PLAZA  
WILMINGTON DE 19894

1 BRIGS COMPANY  
J BACKOFEN  
2668 PETERBOROUGH ST  
HERNDON VA 22071-2443

1 ZERNOW TECHNICAL SERVICES  
L ZERNOW  
425 W BONITA AVE STE 208  
SAN DIMAS CA 91773

1 GENERAL DYNAMICS OTS  
L WHITMORE  
10101 NINTH ST NORTH  
ST PETERSBURG FL 33702

2 GENERAL DYNAMICS OTS  
FLINCHBAUGH DIV  
K LINDE  
T LYNCH  
PO BOX 127  
RED LION PA 17356

1 GKN WESTLAND AEROSPACE  
D OLDS  
450 MURDOCK AVE  
MERIDEN CT 06450-8324

2 BOEING ROTORCRAFT  
P MINGURT  
P HANDEL  
800 B PUTNAM BLVD  
WALLINGFORD PA 19086

NO. OF  
COPIES ORGANIZATION

5 SIKORSKY AIRCRAFT  
G JACARUSO  
T CARSTENSAN  
B KAY  
S GARBO MS S330A  
J ADELMANN  
6900 MAIN ST  
PO BOX 9729  
STRATFORD CT 06497-9729

1 AEROSPACE CORP  
G HAWKINS M4 945  
2350 E EL SEGUNDO BLVD  
EL SEGUNDO CA 90245

2 CYTEC FIBERITE  
M LIN  
W WEB  
1440 N KRAEMER BLVD  
ANAHEIM CA 92806

2 UDLP  
G THOMAS  
M MACLEAN  
PO BOX 58123  
SANTA CLARA CA 95052

1 UDLP WARREN OFC  
A LEE  
31201 CHICAGO RD SOUTH  
SUITE B102  
WARREN MI 48093

2 UDLP  
R BRYNSVOLD  
P JANKE MS 170  
4800 EAST RIVER RD  
MINNEAPOLIS MN 55421-1498

1 LOCKHEED MARTIN  
SKUNK WORKS  
D FORTNEY  
1011 LOCKHEED WAY  
PALMDALE CA 93599-2502

1 LOCKHEED MARTIN  
R FIELDS  
5537 PGA BLVD  
SUITE 4516  
ORLANDO FL 32839

NO. OF  
COPIES ORGANIZATION

1 NORTHROP GRUMMAN CORP  
ELECTRONIC SENSORS  
& SYSTEMS DIV  
E SCHOCH MS V 16  
1745A W NURSERY RD  
LINTHICUM MD 21090

1 GDLS DIVISION  
D BARTLE  
PO BOX 1901  
WARREN MI 48090

2 GDLS  
D REES  
M PASIK  
PO BOX 2074  
WARREN MI 48090-2074

1 GDLS  
MUSKEGON OPER  
M SOIMAR  
76 GETTY ST  
MUSKEGON MI 49442

1 GENERAL DYNAMICS  
AMPHIBIOUS SYS  
SURVIVABILITY LEAD  
G WALKER  
991 ANNAPOLIS WAY  
WOODBIDGE VA 22191

6 INST FOR ADVANCED  
TECH  
H FAIR  
I MCNAB  
P SULLIVAN  
S BLESS  
W REINECKE  
C PERSAD  
3925 W BRAKER LN STE 400  
AUSTIN TX 78759-5316

1 ARROW TECH ASSOC  
1233 SHELBURNE RD STE D8  
SOUTH BURLINGTON VT  
05403-7700

1 R EICHELBERGER  
CONSULTANT  
409 W CATHERINE ST  
BEL AIR MD 21014-3613

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	SAIC G CHRYSSOMALLIS 8500 NORMANDALE LAKE BLVD SUITE 1610 BLOOMINGTON MN 55437-3828
1	UCLA MANE DEPT ENGR IV H T HAHN LOS ANGELES CA 90024-1597
2	UNIV OF DAYTON RESEARCH INST R Y KIM A K ROY 300 COLLEGE PARK AVE DAYTON OH 45469-0168
1	UMASS LOWELL PLASTICS DEPT N SCHOTT 1 UNIVERSITY AVE LOWELL MA 01854
1	IIT RESEARCH CTR D ROSE 201 MILL ST ROME NY 13440-6916
1	GA TECH RESEARCH INST GA INST OF TCHNLGY P FRIEDERICH ATLANTA GA 30392
1	MICHIGAN ST UNIV MSM DEPT R AVERILL 3515 EB EAST LANSING MI 48824-1226
1	UNIV OF WYOMING D ADAMS PO BOX 3295 LARAMIE WY 82071
1	PENN STATE UNIV R S ENGEL 245 HAMMOND BLDG UNIVERSITY PARK PA 16801

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
2	PENN STATE UNIV R MCNITT C BAKIS 212 EARTH ENGR SCIENCES BLDG UNIVERSITY PARK PA 16802
1	PURDUE UNIV SCHOOL OF AERO & ASTRO C T SUN W LAFAYETTE IN 47907-1282
1	STANFORD UNIV DEPT OF AERONAUTICS & AEROBALLISTICS S TSAI DURANT BLDG STANFORD CA 94305
1	UNIV OF MAINE ADV STR & COMP LAB R LOPEZ ANIDO 5793 AEWB BLDG ORONO ME 04469-5793
1	JOHNS HOPKINS UNIV APPLIED PHYSICS LAB P WIENHOLD 11100 JOHNS HOPKINS RD LAUREL MD 20723-6099
1	UNIV OF DAYTON J M WHITNEY COLLEGE PARK AVE DAYTON OH 45469-0240
1	NORTH CAROLINA ST UNIV CIVIL ENGINEERING DEPT W RASDORF PO BOX 7908 RALEIGH NC 27696-7908
5	UNIV OF DELAWARE CTR FOR COMPOSITE MTRLS J GILLESPIE M SANTARE S YARLAGADDA S ADVANI D HEIDER 201 SPENCER LAB NEWARK DE 19716

NO. OF  
COPIES ORGANIZATION

1 DEPT OF MTRL  
SCIENCE & ENGRG  
UNIV OF ILLINOIS  
AT URBANA CHAMPAIGN  
JECONOMY  
1304 WEST GREEN ST 115B  
URBANA IL 61801

1 UNIV OF MARYLAND  
DEPT OF AEROSPACE ENGRG  
A J VIZZINI  
COLLEGE PARK MD 20742

1 DREXEL UNIV  
A S D WANG  
3141 CHESTNUT ST  
PHILADELPHIA PA 19104

3 UNIV OF TEXAS AT AUSTIN  
CTR FOR ELECTROMECHANICS  
J PRICE  
A WALLS  
J KITZMILLER  
10100 BURNET RD  
AUSTIN TX 78758-4497

3 VA POLYTECHNICAL  
INST & STATE UNIV  
DEPT OF ESM  
M W HYER  
K REIFSNIDER  
R JONES  
BLACKSBURG VA 24061-0219

1 SOUTHWEST RESEARCH INST  
ENGR & MATL SCIENCES DIV  
J RIEGEL  
6220 CULEBRA RD  
PO DRAWER 28510  
SAN ANTONIO TX 78228-0510

1 BATELLE NATICK OPERS  
B HALPIN  
313 SPEEN ST  
NATICK MA 01760

3 DIRECTOR  
US ARMY RESEARCH LAB  
AMSRD ARL WM MB  
A FRYDMAN  
2800 POWDER MILL RD  
ADELPHI MD 20783-1197

NO. OF  
COPIES ORGANIZATION

ABERDEEN PROVING GROUND

1 US ARMY ATC  
CSTE DTC AT AC I  
W C FRAZER  
400 COLLERAN RD  
APG MD 21005-5059

91 DIR USARL  
AMSRD ARL CI  
AMSRD ARL O AP EG  
M ADAMSON  
AMSRD ARL SL BA  
AMSRD ARL SL BB  
D BELY  
AMSRD ARL WM  
J SMITH  
H WALLACE  
AMSRD ARL WM B  
A HORST  
T KOGLER  
AMSRD ARL WM BA  
D LYON  
AMSRD ARL WM BC  
J NEWILL  
P PLOSTINS  
A ZIELINSKI  
AMSRD ARL WM BD  
P CONROY  
B FORCH  
M LEADORE  
C LEVERITT  
R LIEB  
R PESCE RODRIGUEZ  
B RICE  
AMSRD ARL WM BF  
S WILKERSON  
AMSRD ARL WM M  
B FINK  
J MCCAULEY  
AMSRD ARL WM MA  
L GHIORSE  
S MCKNIGHT  
E WETZEL  
AMSRD ARL WM MB  
J BENDER  
T BOGETTI  
L BURTON  
R CARTER  
K CHO  
W DE ROSSET  
G DEWING  
R DOWDING  
W DRYSDALE

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
	R EMERSON
	D HENRY
	D HOPKINS
	R KASTE
	L KECSKES
	M MINNICINO
	B POWERS
	D SNOHA
	J SOUTH
	M STAKER
	J SWAB
	J TZENG
	AMSRD ARL WM MC
	J BEATTY
	R BOSSOLI
	E CHIN
	S CORNELISON
	D GRANVILLE
	B HART
	J LASALVIA
	J MONTGOMERY
	F PIERCE
	E RIGAS
	W SPURGEON
	AMSRD ARL WM MD
	B CHEESEMAN
	P DEHMER
	R DOOLEY
	G GAZONAS
	S GHIORSE
	C HOPPEL
	M KLUSEWITZ
	W ROY
	J SANDS
	D SPAGNUOLO
	S WALSH
	S WOLF
	AMSRD ARL WM RP
	J BORNSTEIN
	C SHOEMAKER
	AMSRD ARL WM T
	B BURNS
	AMSRD ARL WM TA
	W BRUCHEY
	M BURKINS
	W GILLICH
	B GOOCH
	T HAVEL
	E HORWATH
	M NORMANDIA
	J RUNYEON
	M ZOLTOSKI

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
	AMSRD ARL WM TB
	P BAKER
	AMSRD ARL WM TC
	R COATES
	AMSRD ARL WM TD
	D DANDEKAR
	T HADUCH
	T MOYNIHAN
	M RAFTENBERG
	S SCHOENFELD
	T WEERASOORIYA
	AMSRD ARL WM TE
	A NIILER
	J POWELL

NO. OF  
COPIES ORGANIZATION

1 LTD  
R MARTIN  
MERL  
TAMWORTH RD  
HERTFORD SG13 7DG  
UK

1 SMC SCOTLAND  
P W LAY  
DERA ROSYTH  
ROSYTH ROYAL DOCKYARD  
DUNFERMLINE FIFE KY 11 2XR  
UK

1 CIVIL AVIATION  
ADMINSTRATION  
T GOTTESMAN  
PO BOX 8  
BEN GURION INTRNL AIRPORT  
LOD 70150  
ISRAEL

1 AEROSPATIALE  
S ANDRE  
A BTE CC RTE MD132  
316 ROUTE DE BAYONNE  
TOULOUSE 31060  
FRANCE

1 DRA FORT HALSTEAD  
P N JONES  
SEVEN OAKS KENT TN 147BP  
UK

1 SWISS FEDERAL ARMAMENTS  
WKS  
W LANZ  
ALLMENDSTRASSE 86  
3602 THUN  
SWITZERLAND

1 DYNAMEC RESEARCH LAB  
AKE PERSSON  
BOX 201  
SE 151 23 SODERTALJE  
SWEDEN

NO. OF  
COPIES ORGANIZATION

1 ISRAEL INST OF TECHLGY  
S BODNER  
FACULTY OF MECHANICAL  
ENGR  
HAIFA 3200  
ISRAEL

1 DSTO  
WEAPONS SYSTEMS DIVISION  
N BURMAN RLLWS  
SALISBURY  
SOUTH AUSTRALIA 5108  
AUSTRALIA

1 DEF RES ESTABLISHMENT  
VALCARTIER  
A DUPUIS  
2459 BLVD PIE XI NORTH  
VALCARTIER QUEBEC  
CANADA  
PO BOX 8800 COURCELETTE  
GOA IRO QUEBEC  
CANADA

1 ECOLE POLYTECH  
J MANSON  
DMX LTC  
CH 1015 LAUSANNE  
SWITZERLAND

1 TNO DEFENSE RESEARCH  
R IJSSELSTEIN  
ACCOUNT DIRECTOR  
R&D ARMEE  
PO BOX 6006  
2600 JA DELFT  
THE NETHERLANDS

2 FOA NATL DEFENSE RESEARCH  
ESTAB  
DIR DEPT OF WEAPONS &  
PROTECTION  
B JANZON  
R HOLMLIN  
S 172 90 STOCKHOLM  
SWEDEN



NO. OF  
COPIES ORGANIZATION

- |   |  |
|---|--|
| 2 | DEFENSE TECH & PROC<br>AGENCY GROUND<br>I CREWTHERR<br>GENERAL HERZOG HAUS<br>3602 THUN<br>SWITZERLAND               |
| 1 | MINISTRY OF DEFENCE<br>RAFAEL<br>ARMAMENT DEVELOPMENT<br>AUTH<br>M MAYSELESS<br>PO BOX 2250<br>HAIFA 31021<br>ISRAEL |
| 1 | TNO DEFENSE RESEARCH<br>I H PASMAN<br>POSTBUS 6006<br>2600 JA DELFT<br>THE NETHERLANDS                               |
| 1 | B HIRSCH<br>TACHKEMONY ST 6<br>NETAMUA 42611<br>ISRAEL   |
| 1 | DEUTSCHE AEROSPACE AG<br>DYNAMICS SYSTEMS<br>M HELD<br>PO BOX 1340<br>D 86523 SCHROBENHAUSEN<br>GERMANY              |

NO. OF  
COPIES ORGANIZATION

ABERDEEN PROVING GROUND

15    DIR USARL  
      AMSRD ARL WM TC  
      L MAGNESS  
      AMSRD ARL WM TB  
      J WATSON  
      AMSRD ARL WM TD  
      S SEGLETES  
      AMSRD ARL WM MB  
      W DE ROSSET (10 CPS)  
      AMSRD ARL WM MC  
      V CHAMPAGNE  
      AMSRD ARL WM MD  
      B SCOTT

INTENTIONALLY LEFT BLANK.